

NON-UNIFORM ONSET OF NUCLEATE FLOW BOILING OF R-134A INSIDE A GLASS MINICHANNEL

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KEYWORDS

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ABSTRACT

Even if far from being completely characterized, convective boiling inside mini- and micro-channels is likely to be one of the most relevant process for enhancing the cooling capability of modern heat transfer devices. Since, due to the process miniaturization, many novel devices are very small, fully developed flows are difficult to achieve, and therefore the knowledge of the onset nucleate boiling (ONB) position is even more critical for cooling design. However visualization of the ONB phenomena is not common in the literature, since usually the heaters are made in opaque metals and therefore they are not allowing a complete imagining. The present work consists in an experimental investigation of the incipient boiling of R134a inside a circular glass mini-channel mounted horizontally and equipped with a series of transparent Indium Tin Oxide (ITO) heaters. The effects of the heat flux input levels and the refrigerant mass fluxes on the ONB process and on the saturated boiling heat transfer rate are quantitatively explored. The flow pattern visualizations, carried on by means of a high-speed camera, show that the nucleation process is oddly non-uniform: the first vapor bubbles are always generated on the upper side of the tube and lead to a first wall temperature drop. A further increase in the heat flux values results in an increased wall superheat until bubbles nucleation originates also on the lower side of the tube causing a second wall temperature drop. Finally, at higher heat input levels, the boiling process becomes uniformly distributed on the inner tube surface. This phenomenon occurred also after a 180° rotation of the glass

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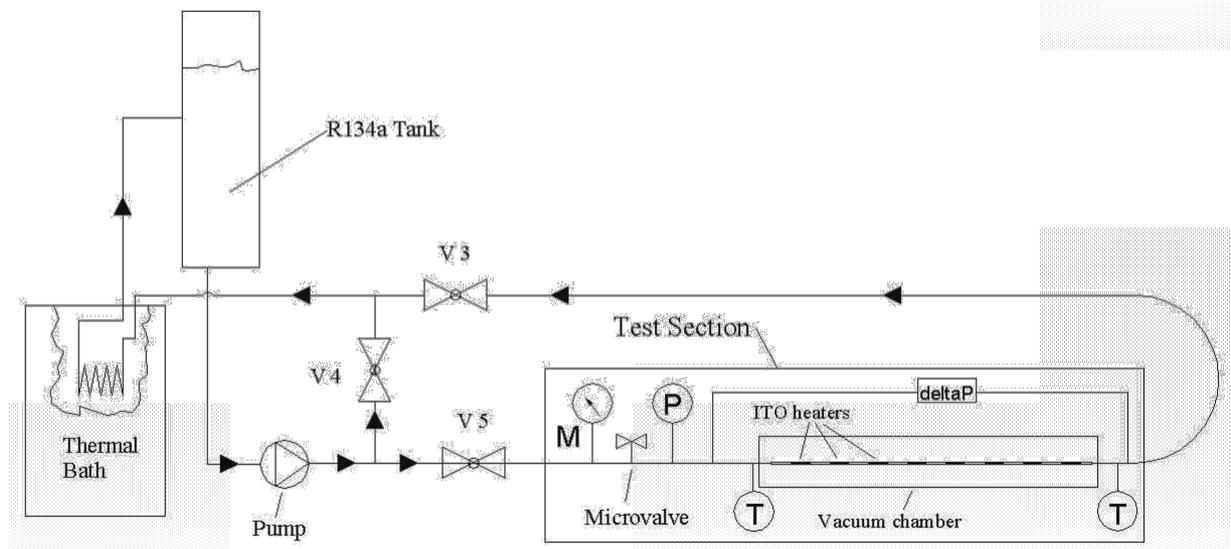
27 tube and, after a critical analysis of the potential origins, it remains presently unexplained. An
28 evaluation of the heat transfer coefficients for low vapor quality regimes is finally presented.

29 1. INTRODUCTION

30 Characterizing two-phase heat transfer phenomena inside mini and micro flow boiling
31 systems is fundamental to understand how to increase the cooling capability of modern heat
32 transfer devices. The need of an increased cooling capability is due to the smaller size of the
33 components and the consequent higher specific heat generation. The complex nature of two-
34 phase flows in mini and micro-channels needs still to be completely understood, as underlined
35 by Thome [1,2], Celata [3], Harirchian and Garimella [4] and recently reviewed by Baldassari
36 and Marengo [5]. After more than 30 years, flow patterns, void fraction measurement, heat
37 transfer coefficient evaluation, onset of nucleate boiling are still open research topics for the
38 flow boiling in mini and microchannels [5,6]. Focusing on the ONB, which marks the
39 boundary between the single-phase and two-phase heat transfer region, Ghiaasiaan et al. [7]
40 developed a semi-empirical method for the estimation of the incipient boiling heat flux. Bang
41 et al. [8] performed a digital photographic study of subcooled flow boiling to observe near-
42 wall structures, giving primary attention to bubble formation and bubble coalescence in the
43 bubble layer. Callizo et al. [9] presented boiling curves for R-134a and a comparison between
44 the experimental heat transfer coefficients and the predictions from classical correlations. Liu
45 et al. [10] investigated experimentally ONB of water in a rectangular micro-channel
46 developing an analytical model able to predict the heat flux and the bubble size at the onset of
47 boiling. Recently, Hong et al. [11] found that the heat flux and wall superheat needed to
48 initiate the nucleate boiling in narrow rectangular channel under static conditions, increased
49 with the mass flux and the inlet subcooling.

50 The present paper focuses on the experimental analysis of the heterogeneous onset of nucleate
51 flow boiling of R-134a in a mini-channel having 4 mm internal diameter, with a particular
52 non symmetrical physical outcome, which is making the present research inspiring further
53 investigations. In fact the use of a transparent glass tube with transparent heaters allows the
54 visualization of the ONB and evidences a gradual activation of the nucleation sites, starting
55 from the upper side of the heater to the lower side, for all the tested mass flux. The
56 corresponding boiling curves evidence two different temperature drops, associated to the
57 starting of the nucleation from the upper side and from the lower side of the heater,
58 respectively. Two temperature drops have been also observed by Piasecka et al. [12] during

59 their boiling investigation. They found a two-stepped phase of the boiling phenomenon,
 60 similar to a second kind hysteresis, characteristic of nucleate pool boiling on developed
 61 micro-surfaces. They could not associate this kind of non-uniform condition to the flow
 62 pattern observation, since their experimental test-rig was not able to capture the bubble
 63 formation. In the present paper, ONB is visualized and recorded by means of a high-speed
 64 camera thanks to the ITO heaters, whose transparency allows a full characterization of the
 65 earliest stages of the bubbles formation. Several boiling curves are presented at different mass
 66 flux levels, $G = 50, 80, 100, 115$ and $137 \text{ kg/m}^2\text{s}$ together with the flow pattern visualization.
 67 Finally the heat transfer coefficient is calculated as a function of the heat flux and of the wall
 68 superheat for each mass flux.



69
 70 **Figure 1 Experimental test rig scheme**

71 2. EXPERIMENTAL SET-UP AND PROCEDURE

72 2.1 Experimental set-up

73 A schematic view of the experimental apparatus is given in Figure 1. The test section tube is
 74 made of high precision glass (DURAN) with an internal diameter of 4 mm. The Eotvos
 75 number $Eo = g(\rho_L - \rho_V)L^2/\sigma$ is hence set to 21.7, i.e. the flow boiling process is studied in
 76 the so-called macroscale region. The transparent ITO film covers eight portions of equal
 77 length (40 mm) and each independent portion is used to electrically heat the glass surface
 78 allowing to observe and record the boiling mechanisms taking place inside the tube with a
 79 high speed PCO® camera. The minichannel is enclosed in a co-axial glass tube with 60mm
 80 external diameter vacuumed in order to eliminate the convection heat losses and consider only

81 radiation losses. The temperature of the tube wall is monitored by means of eight K-type
82 thermocouples ($\pm 0.2^\circ\text{C}$ after calibration) placed externally on each heater at the upstream side
83 and at the middle of the tube height, as represented in Figure 2.
84

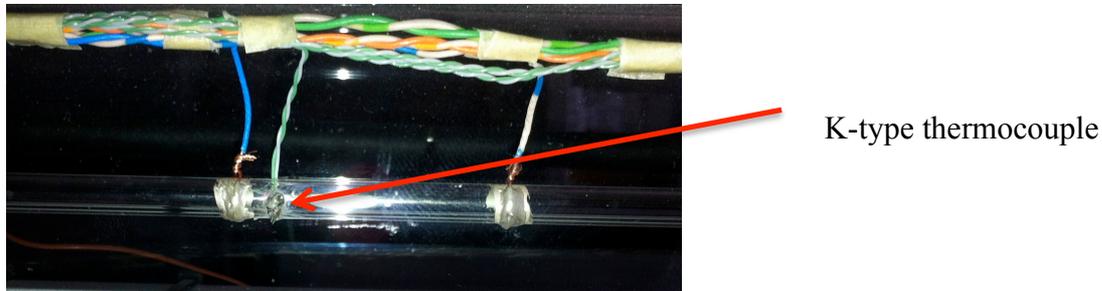


Figure 2 ITO heater and the position of the K-type thermocouple

85
86 During this experimental campaign one heater with a heat flux up to 30 kW/m^2 has been used.
87 Refrigerant temperature is measured both at the test section inlet and outlet by two K-type
88 thermocouples; the pressure is measured at the inlet by a PSE 510 pressure sensor.
89 The test rig consists of two main circuits: the test refrigerant loop and a secondary loop, where
90 the fluid is thermally controlled. The loop is filled with R134a until the internal pressure
91 reaches 6 bar, in order to bring the saturation temperature just below the ambient temperature.
92 A helix type heat exchanger is cooled by means of a thermal bath HAAKE C50P[®]; this unit is
93 necessary to define the operating experiment temperature and to condense the fluid coming
94 from the test section in the form of wet vapor. The mass flow rate is measured by a Coriolis
95 flow meter CORI-FLOW[®]. ITO coatings power is controlled by the PC through the NI
96 (National Instruments) USB-6008[®] acquisition and control device and by the specifically
97 designed electronic board, which is supplied with a 56V DC power supply. The electronic
98 board applies the correct value of voltage to the heater in order to reach the desired electrical
99 power and amplifies the coating voltage and current signals so that these can be acquired by
100 the data acquisition module and transmitted to the PC. The radiation heat transfer between the
101 ITO coating and the environment is numerically estimated and subtracted to the total power
102 supplied to the ITO coatings in order to obtain the net thermal power supplied to the working
103 fluid; this value is controlled with a standard PID (proportional-integral-derivative).
104 Temperature and pressure signals are acquired using Agilent 34970A[®]. The system control,
105 data acquisition software has been developed with Simulink[®] and dedicated MATLAB[®]

106 programs have been developed for the data post processing. Other details on the test rig are
107 given in [13].

108 **2.2 Experimental procedure**

109 Before running the experiments, the coaxial glass chamber is vacuumed, the mass flux is
110 adjusted to the desired value, the fluid inlet temperature is set in order to enter the test section
111 in sub-cooled condition of maximum 2K, and the system works as close as possible to the
112 saturated boiling conditions. The heater power supply is set to the desired value which is
113 maintained until a steady tube temperature is reached. Mass flux, temperature, pressure and
114 power input values are stored using the data acquisition system. In order to define the
115 characteristic boiling curves, the power supplied to the fluid is increased, starting from 0 W,
116 with increments of 0.5W in the regions near the onset of nucleate boiling, and of 1 W in the
117 other regions until the maximum power ($Q_{\max} = 11$ W). Then the power level is decreased
118 from the maximum value to 0 W with step of 1 W. For each heat flux step, the corresponding
119 flow pattern is visualized and recorded using the high-speed camera. In order to provide the
120 necessary amount of light, a incandescent lamp is positioned in front of the camera. Because
121 the lamp temperature is very high, a glass covered with an infrared filtering film has been
122 located between the lamp and the test section, in order to minimize the radiated power.

123 The experimental procedure explained above is repeated two times for every mass flux value
124 examined, namely $G = 50$ kg/m²s, $G = 80$ kg/m²s, $G = 100$ kg/m²s, $G = 115$ kg/m²s and $G =$
125 137 kg/m²s in order to check the repeatability. Finally the temperature difference, usually
126 called wall superheat, is calculated for each heat input level:

$$127 \quad \Delta T_{sh} = T_{w,in} - T_{sat} \quad (1)$$

128 where $T_{w,in}$ is the internal temperature of the heated wall, and it is calculated starting from the
129 temperature of the external surface of the mini-channel measured by the single K-type
130 thermocouple in Figure 2, considering the heat conduction through the glass. T_{sat} is the fluid
131 inlet saturation temperature estimated with the help of the NIST REFPROP libraries [14],
132 knowing the fluid inlet pressure measured by the pressure sensor. Regarding the experimental
133 errors, the mass flow rate accuracy is $\pm 1\%$ while the pressure accuracy is $\pm 0.5\%$ in the
134 pressure range of R-134a experiments (≈ 5.6 bar). The thermocouples have an accuracy of \pm
135 0.2°C after calibration.

136 **3. EXPERIMENTAL RESULTS**

137 Thanks to the transparency of the ITO heaters, it is possible to visualize the earliest stages of
 138 the bubbles formation and the phase transition phenomena (Figure 3 and 4). The flow patterns
 139 visualization reveals that the first nucleation event always appears on the upper side of the
 140 heated tube for all the mass fluxes tested. Only after a further increase of the heat flux, the
 141 nucleation starts also on the lower side of the heated tube. When the nucleation appears, it is
 142 associated to a temperature drop. Hence the boiling curves presented in the next section
 143 evidence two temperature drops, that are respectively associated to the upper side and to the
 144 lower side onset of nucleate boiling.

145 The dimensionless numbers relative to the experimental conditions of Table 1 are given in
 146 Table 2.

Refrigerant	Inlet pressure [MPa]	Tsat [°C]	G range [kg/m ² s]	q'' [kW/m ²]	x [-]
R-134a	0.537-0.590	18-21	50-137	0-22	max 0.093

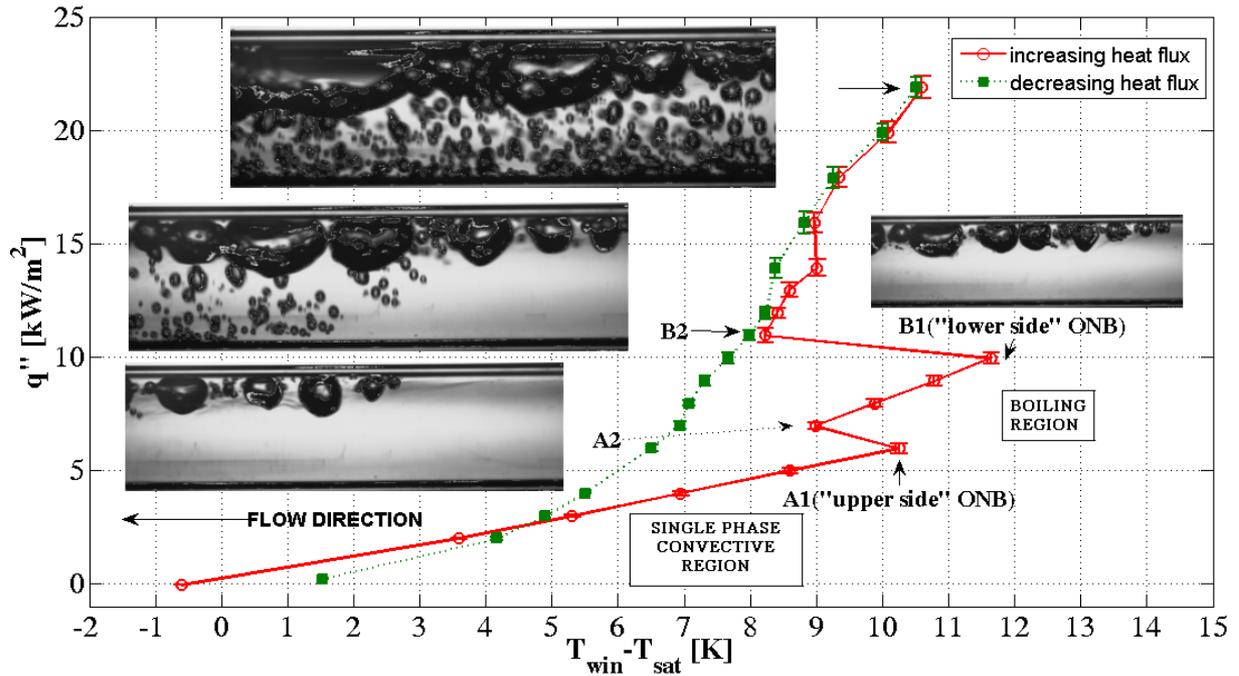
147 **Table 1 Summary of the experimental conditions**

Specific mass flow rate	Liquid phase Capillary number	Liquid phase Reynolds number	Vapor phase Reynolds number	Boiling number	Confinement number
G	Ca(LO)	Re(LO)	Re(VO)	Bl	Cn
[kg/m ² s]	$Ca_{LO} = \frac{\mu_L G}{\rho_L \sigma}$	$Re_{LO} = \frac{G d_h}{\mu_L}$	$Re_{VO} = \frac{G d_h}{\mu_V}$	$Bl = \frac{q''}{G h_{fg}}$	$Cn = \left[\frac{1-x}{x} \right]^{0.9} \cdot \left[\frac{\rho_V}{\rho_L} \right]^{0.5}$
50	0.00097	975	17357	0.0024	1.161
80	0.00156	1560	27771	0.0015	1.479
100	0.00194	1949	34713	0.0012	1.803
115	0.00223	2242	39920	0.0011	2.155
137	0.00266	2671	47557	0.0009	3.023

148 **Table 2 The dimensionless number values for the experimental conditions**

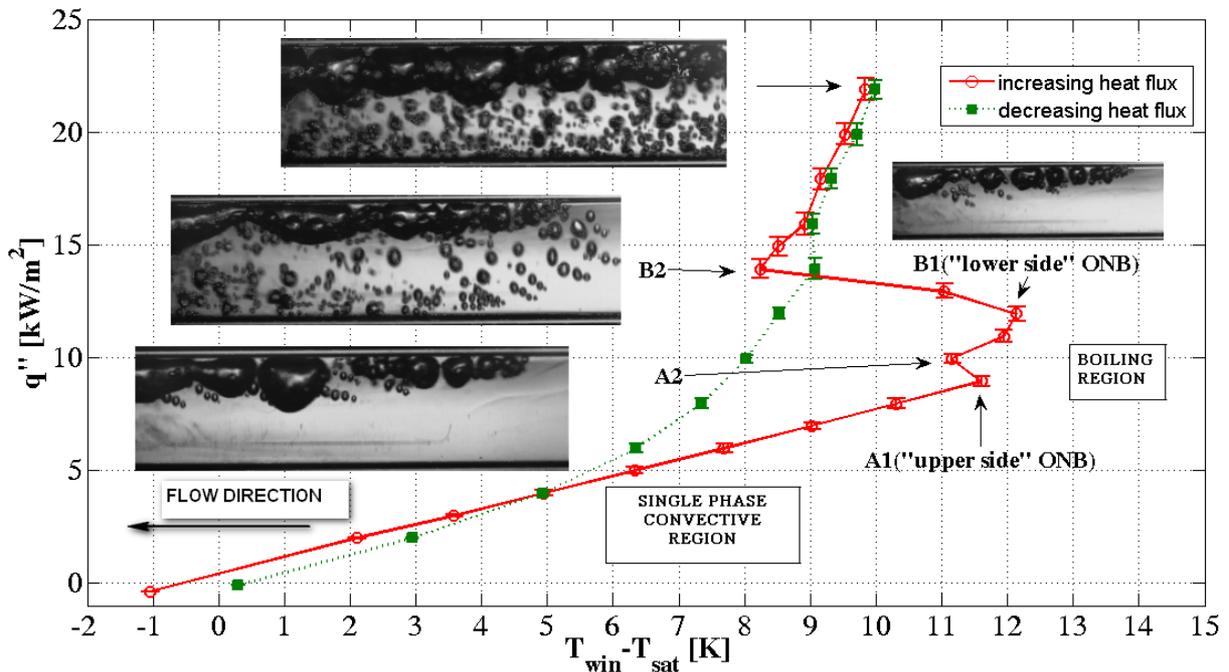
149 3.1 Boiling curves and flow patterns

150 The heat flux versus ΔT_{sh} represents the boiling curve obtained during the experiments. In
 151 Figure 3 and 4 the boiling curves and the flow patterns corresponding to $G = 115 \text{ kg/m}^2\text{s}$ and
 152 $G = 137 \text{ kg/m}^2\text{s}$ are presented. The red line corresponds to the increasing in the heat flux,
 153 while the green line to the heat power decreasing. The boiling curves and the flow patterns
 154 corresponding to the other mass fluxes tested in this paper are presented in [15].



155

156 **Figure 3 Boiling curve and flow patterns associated to points A2, B1, B2 and to $q''=22$**
 157 **kW/m^2 for $G=115 \text{ kg/m}^2\text{s}$ and a maximum vapor quality of 0.04.**



158

159 **Figure 4 Boiling curve and flow patterns associated to points A2, B1, B2 and to $q''=22$**
 160 **kW/m^2 for $G=137 \text{ kg/m}^2\text{s}$ and a maximum vapor quality of 0.035.**

161 As usual, the values of wall superheat during the increasing of the heat flux are different from
 162 those obtained during the heat flux reduction (“boiling hysteresis”). Two main regions,

163 namely *single phase forced convective region* and *boiling flow region*, are clearly identified in
164 the figures above. Regarding the red lines in Figure 3 and 4, the heat flux increases almost
165 linearly with the increasing in the wall superheat in the single phase convective region. The
166 liquid refrigerant in contact with the tube internal surface soon becomes superheated, whereas
167 the fluid bulk may remain saturated or even slightly subcooled. Further increase in the heat
168 flux results in an increased wall superheat; vapor nuclei are activated and the boiling process
169 occurs in the upper side of the tube. The ONB heat transfer mechanism causes the heating
170 surface temperature to drop, resulting in a reduction of the wall temperature (A2 in Figure 3
171 and 4). A further increase of the heat flux results in an increase of the number of bubble sites
172 and the wall superheat is again slowly increasing.

173 In our experiment however a second temperature drop is clearly recognizable. From the
174 visualizations it is possible to enlighten that this second temperature drop occurs when the
175 nucleation starts also in the lower side of the heater. Hence such temperature drop (B2 in
176 Figure 3 and 4) is due to a heterogeneous but non-uniform flow boiling process.

177 In order to further clarify the nomenclature A1, A2, B1, B2 used in the figures above, it is
178 important to define:

- 179 - A1 as the “upper side” ONB point; it corresponds to the maximum heat flux value at which
180 only liquid phase exists;
- 181 - A2 corresponds to the maximum temperature drop occurring after A1. Then a further
182 increase of the heat input level results in an increase of wall superheat;
- 183 - B1 is the “lower side” ONB, defined as the maximum wall superheat below which boiling is
184 still only activated in the upper side of the heater.
- 185 - B2 corresponds to the maximum temperature drop occurring after B1.

186 In B2 the nucleation starts also in the lower part of the heater. Afterwards boiling occurs more
187 uniformly on both the upper and lower tube surfaces, and the red curve is showing a more
188 regular trend, since the wall superheat increases almost linearly with the heat flux.

189 **3.3 Hypothesis on the origins of non-uniform onset nucleate boiling**

190 Four hypotheses were made by the authors to explain why in our experiment the nucleation
191 always starts in the upper side of the heater:

- 192 - the hydrostatic difference of pressure between the upper side and the lower side of the heater
193 could cause a difference in saturation temperature, which could be responsible for the non-
194 uniform boiling;
- 195 - a different surface roughness of the glass tube could influence the nucleation process;

196 - the ITO coating could have been spattered with a non homogeneous thickness.
197 We have then achieved the following answers:
198 - the hydrostatic difference of pressure in a channel having 4 mm internal diameter at 20°C is
199 forty times lower than the pressure needed to increase the saturation temperature of 1°C;
200 - the surface roughness of the glass is in the order of the nanometers and no scratch or defect
201 is optically noticeable on the surface at least at the microscale;
202 - the “upper side” and the “lower side” ONB were observed repeating the experiment using all
203 eight different ITO heaters in different positions along the tube;
204 - since the plasma coating procedure can originate the same non-uniformity for all the 8
205 heaters, the experiment was then repeated after rotating the tube of 180°. It was confirmed
206 that the boiling always starts in the upper side of the heater.
207 Additionally the low thermal diffusivity of the glass, used for the tubes, is of course
208 amplifying the effects of the non-uniform distribution of nucleation sites, since the internal
209 wall conduction is not able to reduce thermal gradients.

210 [A possible temperature gradient inside the refrigerant flow could be a reason for such](#)
211 [phenomenon, even if the inlet of the minichannel is at least 600mm \(150 D_i\) far from the last](#)
212 [geometrical change of the tube, which occurs just after the Coriolis flow meter, named M in](#)
213 [Figure 1.](#)

214 A further possibility to explore is that, because of the presence of gas inside the refrigerant, a
215 number of gas molecules will accumulate in form of nanobubbles on the inner tube surface
216 [16,17,18]. The presence of gas inside the refrigerant, according to the standard Ahri 700-
217 2011, can be estimated as 1,5% vol at 298 K. Due to buoyancy forces the number of the
218 nanobubbles will be higher on the top side with respect to the bottom side. These pre-existing
219 air embryos entrapped in the flow and concentrated in the upper side of the mini-channel
220 could be the responsible for the starting of the nucleation on the upper side.

221 3.4 Heat transfer coefficients

222 From each boiling curve obtained decreasing the heat flux, it was possible to calculate the
223 heat transfer coefficient according to the equation:

$$225 \quad q'' = h (T_{\text{win}} - T_{\text{sat}}) \quad (2)$$

226
227 In Figures 5-7 the heat transfer coefficients are presented with the error bars; the uncertainties
228 are calculated according to the [theory of error propagation, starting from the errors on the heat](#)

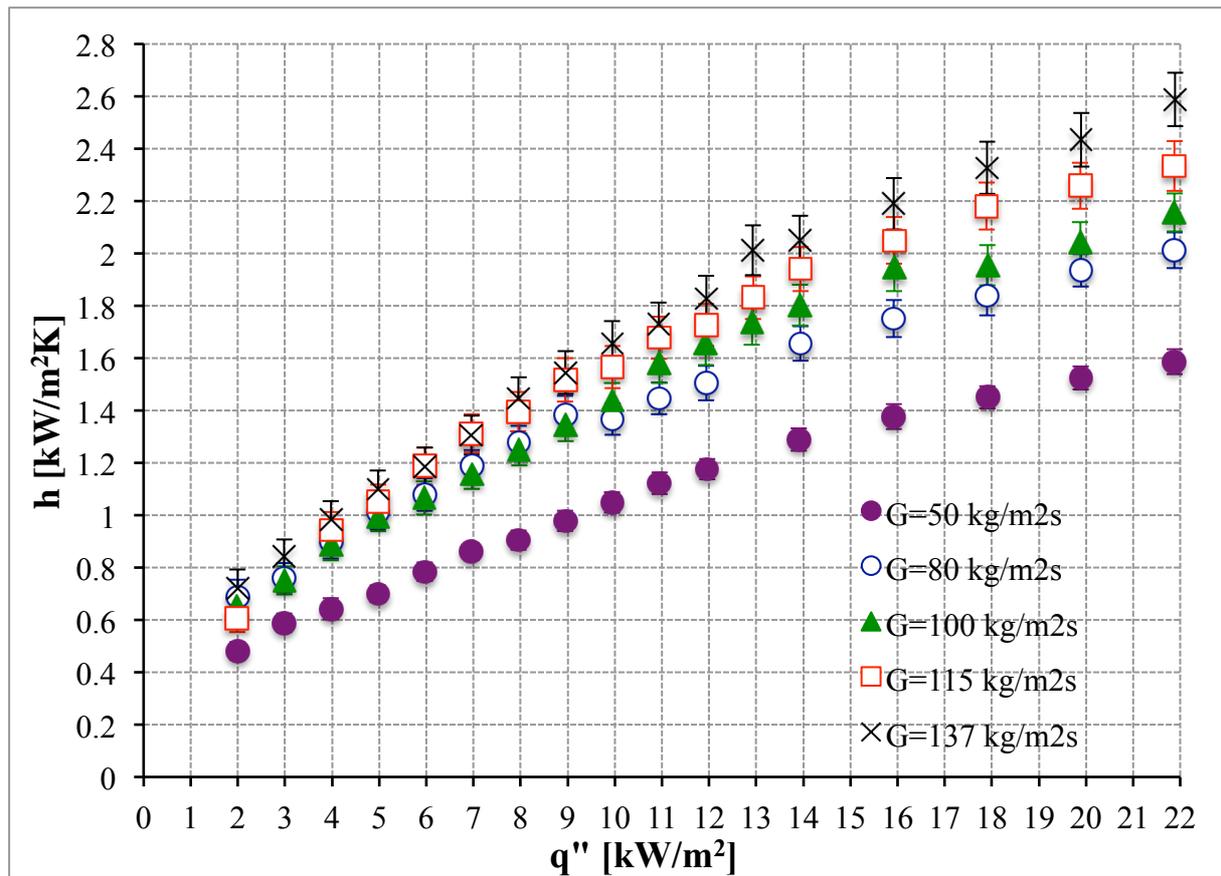
229 flux and on the wall superheat. In Table 3 the sensor accuracy and the estimated uncertainties
 230 of the derived quantities are summarized.

231

Sensors accuracy	Mass flow rate	Pressure (at ≈ 5.6 bar)	K-type thermocouples (after calibration)
	$\pm 1\%$	$\pm 0.5\%$	$\pm 0.2^\circ\text{C}$
Maximum estimated uncertainty	Heat flux	Wall superheat	Heat transfer coefficient
	$\pm 4\%$	$\pm 7\%$	$\pm 8\%$

232 **Table 3 Summary of the sensor accuracy and the estimated uncertainties of the derived**
 233 **quantities**

234 Boiling curves and the associated flow patterns for $G = 50, 80$ and $100 \text{ kg/m}^2\text{s}$ are presented
 235 in [15]. It is important to point out that the data obtained for $G = 50 \text{ kg/m}^2\text{s}$ have the
 236 peculiarity that phase A1 may not correspond only to a single liquid phase, since first
 237 nucleation events may occur even before, without a measurable variation of the wall
 238 superheat. This effect is due to the fact that a single thermocouple for each heater is not able
 239 to capture the local temperature variation around the bubble nucleation site. Such
 240 measurement weakness may produce larger experimental inaccuracies, even with the current
 241 high precision, especially for low wall superheat temperatures.

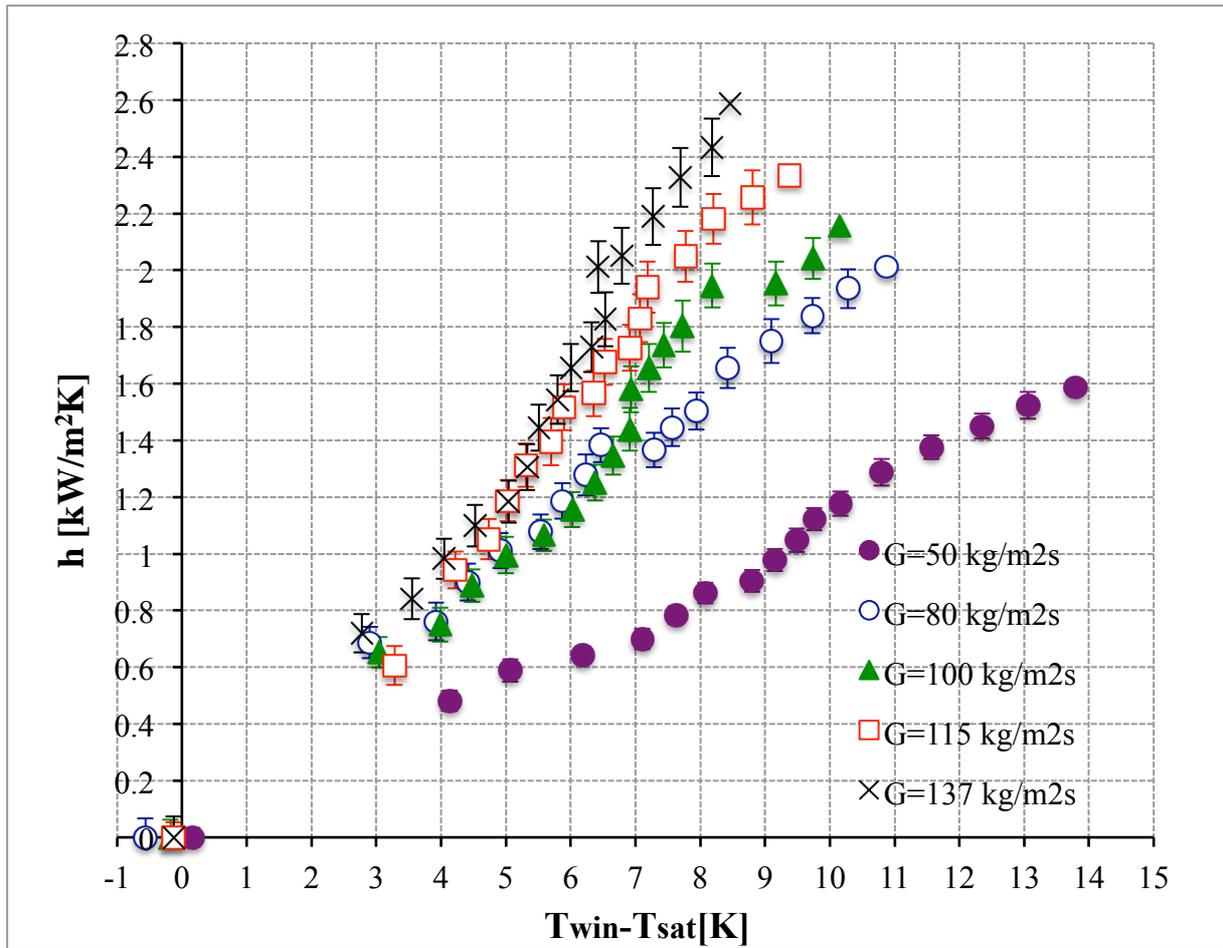


242

243 **Figure 5** The heat transfer coefficient h , evaluated from each boiling curve - obtained
 244 decreasing the heat flux - as function of the heat flux.

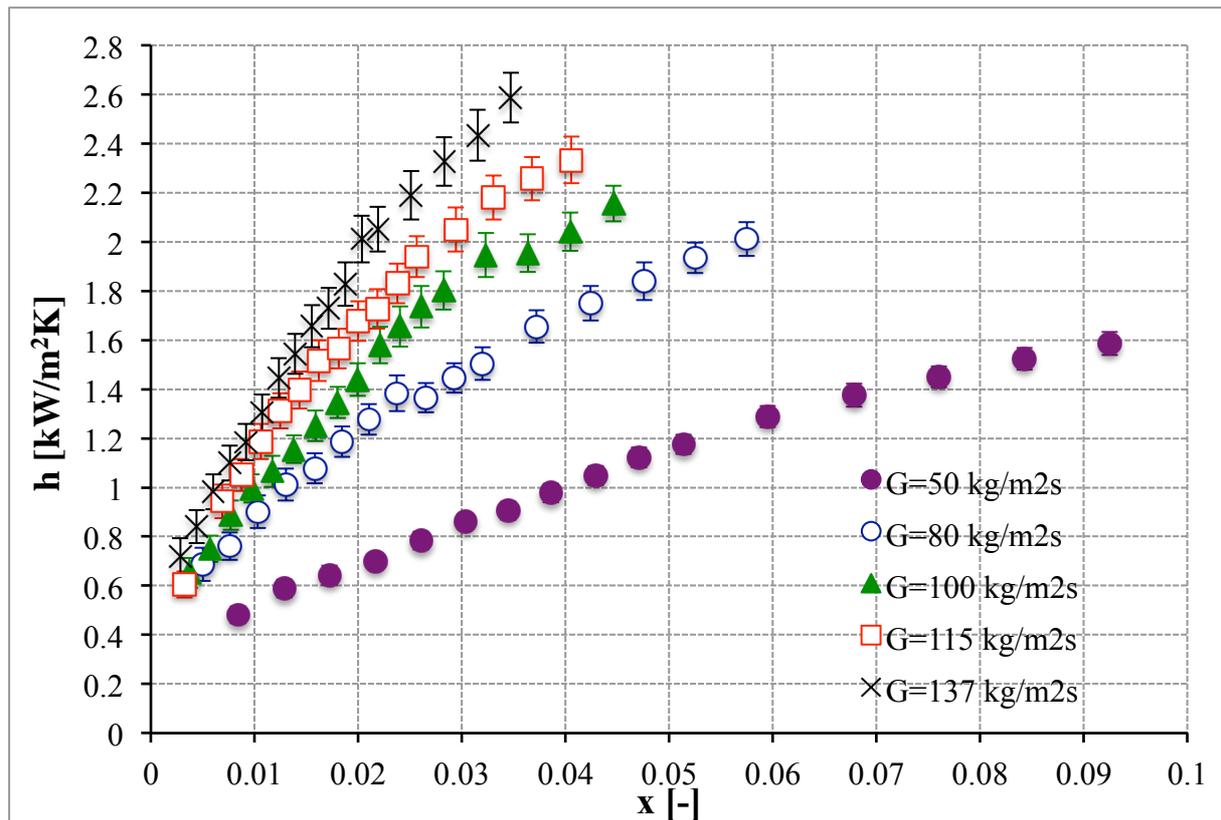
245 In Figure 5 for a heat flux about 9 kW/m^2 the data obtained with $G=100 \text{ kg/m}^2\text{s}$ apparently
 246 decrease more than the data with $G=80 \text{ kg/m}^2\text{s}$, keeping lower values until the minimum heat
 247 flux of 2 kW/m^2 is reached. Such a trend inversion is explained considering only the
 248 experimental accuracy, since there is no relevant change in the flow patterns.

249



250

251 **Figure 6** The heat transfer coefficient h , calculated from each boiling curve - obtained
 252 decreasing the heat flux - as function of the wall superheat.

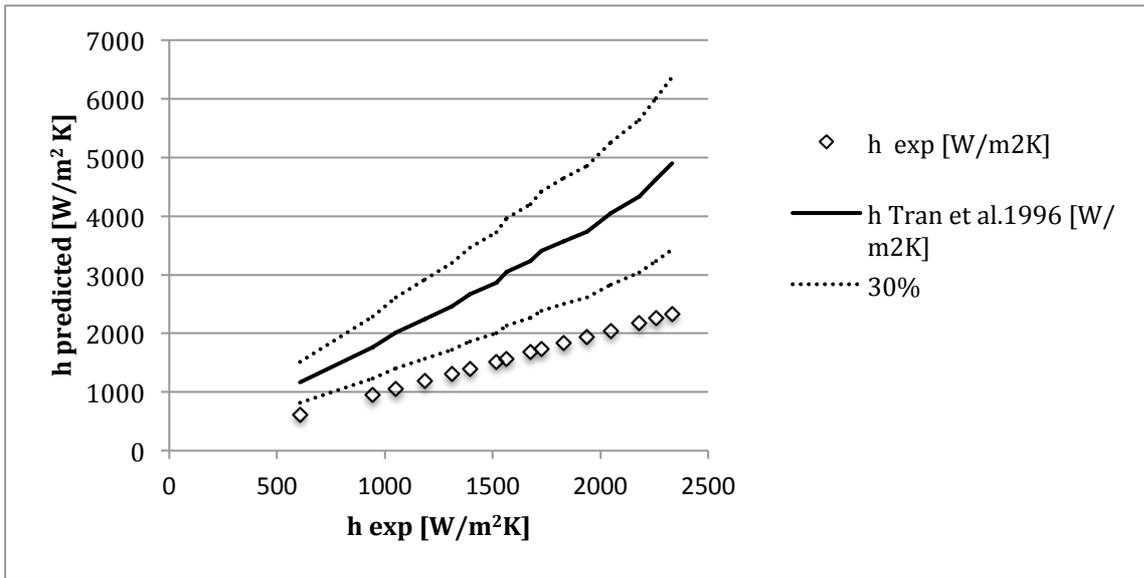


253

254 **Figure 7 The heat transfer coefficient h , calculated from each boiling curve - obtained**
 255 **decreasing the heat flux - as function of the vapor quality.**

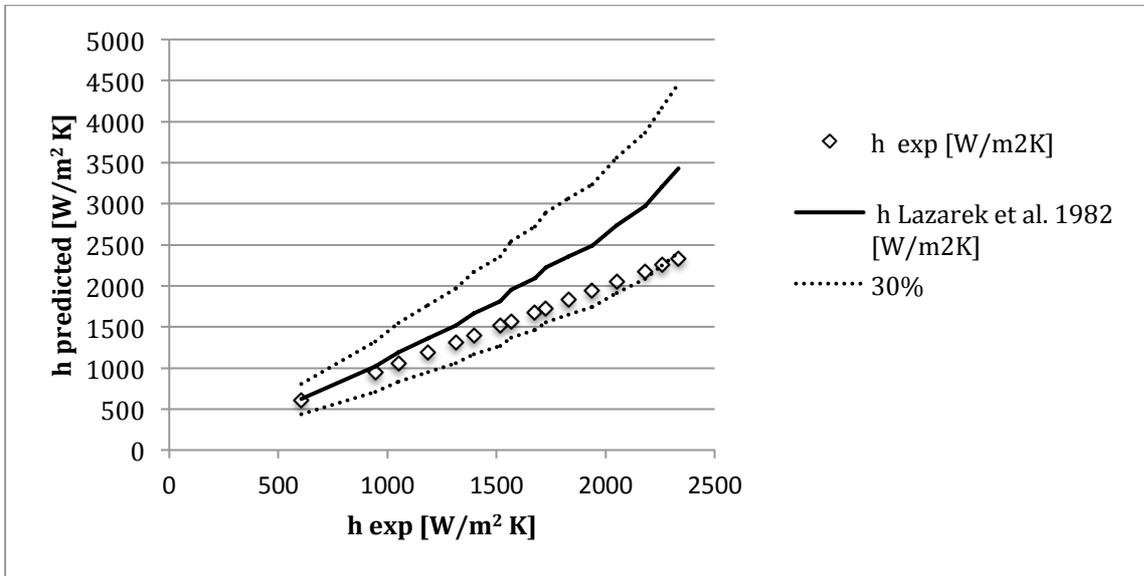
256 The heat transfer coefficient, estimated at one fixed heat flux value or one wall superheat
 257 value, increases with the mass flux level. For such very low vapor qualities, the heat transfer
 258 coefficient increases with the vapor quality, meaning that more active bubbling (in terms of
 259 frequency and nucleation site density) is increasing the heat transfer from the wall. Note that
 260 this appears to be in contrast with many experimental studies, among which [19,20,21], which
 261 predict a decrease of the heat transfer coefficient with vapor quality.

262 A final important consideration is related to the comparison of the obtained heat transfer
 263 coefficient values with some common correlations of the literature, summarized in table 4,
 264 which are given in Figure 8 a,b,c.



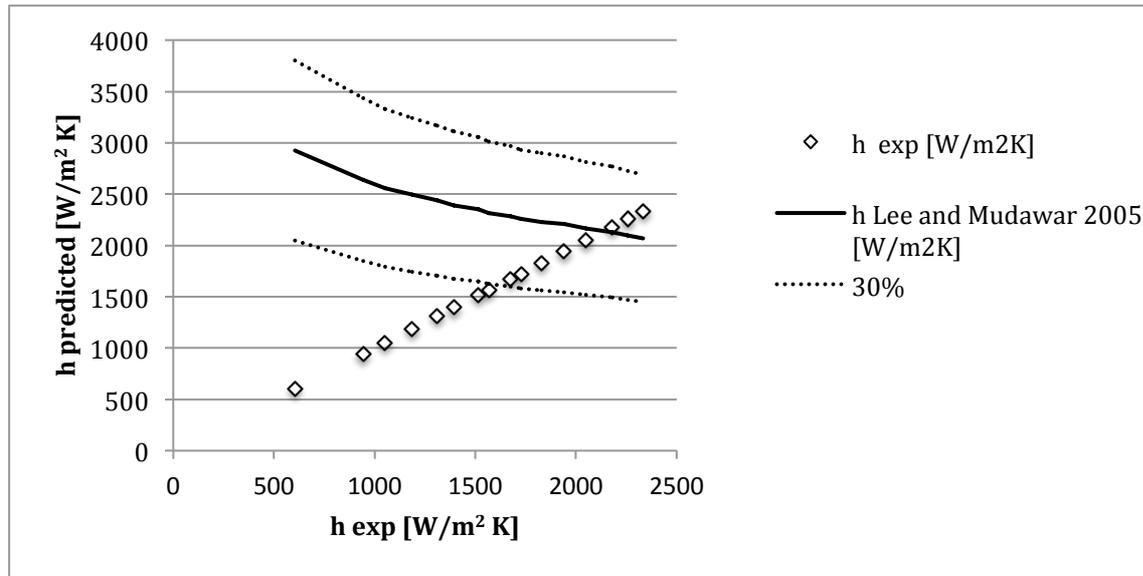
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266 (a)



267

268 (b)



269

270 (c)

271 **Figure 8 Comparison of the heat transfer coefficients, calculated from the experimental**
 272 **boiling curve decreasing the heat flux ($G=115\text{kg/m}^2\text{s}$, vapor quality range $0<x<0.04$,**
 273 **open symbols) with standard correlations in literature summarized in Table 4 (lines).**

Name of the correlation	Equation and experimental range
Lazarek and Black [23]	$h = (30 \text{Re}_{LO}^{0.857} \text{Bl}^{0.714}) \frac{k_L}{d_h} \text{ R-113}$ <p>$14 < q'' < 380 \text{ kW/m}^2$ $125 < G < 750 \text{ kg/m}^2\text{s}$ $0 < x < 0.6$</p>
Tran et al. [22]	$h = 8.4 \cdot 10^5 \cdot (We_{LO} \text{Bl}^2)^{0.3} \left(\frac{\rho_L}{\rho_V} \right)^{-0.4} \text{ R-12}$ <p>$3.6 < q'' < 129 \text{ kW/m}^2$ $44 < G < 832 \text{ kg/m}^2\text{s}$ $0 < x < 0.94$</p>
Lee and Mudawar [24]	$h = 3.856 X^{0.267} h_L \text{ for } x < 0.05$ <p>R-134a, $159 < q'' < 938 \text{ kW/m}^2$ $127 < G < 654 \text{ kg/m}^2\text{s}$</p>

274 **Table 4 The empirical correlations of Lazarek and Black [23], Tran et al [22] and Lee**
 275 **and Mudawar [24]**

276 The correlations of Lazarek and Black [23] and Tran et al. [22] do not contain the vapor
 277 quality parameter. Therefore, since the present experiments have been done in a narrow range
 278 of vapor qualities, the correlation of Lee and Mudawar [24] was also considered. They
 279 proposed a correlation for the heat transfer coefficient for R-134a and water in three ranges of

280 vapor quality: $x < 0.05$, $0.05 < x < 0.55$, $0.55 < x < 1$. The correlation for vapor quality lower than
281 0.05 was implemented in the comparison even if the heat flux range for such correlation is
282 higher than 22 kW/m^2 (Figure 8). The heat transfer coefficient h_L for the liquid phase was
283 calculated according to Shah correlation [25].

284 The measured heat transfer coefficient values are in good agreement with the correlation of
285 Lazarek et al. [22], with larger deviations with respect to the correlation of Tran et al. [23],
286 while a strong departure with the correlation of Lee and Mudawar [24] appears. This
287 correlation, for vapor quality lower than 0.05, was in fact validated only for water and none of
288 the R-134a data. For a heat flux up to 22 kW/m^2 , the values of the heat transfer coefficients
289 are strictly related to the increase of bubble number and frequency. The estimated values of
290 the heat transfer coefficient appear to be lower than expected, interpolating data for higher
291 vapor quality values. The use of a single thermocouple on the minichannel and the fact that
292 the glass has a low thermal conductivity, partially motivate the under-evaluation of the heat
293 transfer coefficient.

294 4. CONCLUSIONS

295 Saturated heterogeneous non-uniform flow boiling of R134a in horizontal mini-channel has
296 been experimentally investigated at five different mass flux levels (50, 80, 100, 115 and 137
297 $\text{kg/m}^2\text{s}$) with particular attention on the visualization of the onset of nucleate boiling. The
298 boiling curves evidence two different temperature drops and this is due to the fact that
299 nucleation always starts in the upper side of the heaters and only increasing the heat flux,
300 activating more nucleation sites, boiling starts later in the lower part of the heater. The flow
301 pattern visualizations highlight this non-uniform boiling. The reason for such behavior is still
302 under investigation and it should be also linked to the low thermal diffusivity of the glass,
303 which is rarely used in flow boiling studies as tube material. The heat transfer coefficient has
304 been estimated as function of the heat flux and of the wall superheat for all tested mass fluxes.
305 It emerges that, as expected, the heat transfer coefficient values, estimated at a fixed heat flux
306 value or wall superheat value, increases with the mass flux level. With such very low vapor
307 quality values, the heat transfer coefficients appear to increase with the vapor quality, i.e. with
308 the number and frequency of bubble detaching from the surface. The estimated values of the
309 heat transfer coefficients are ranging from $500 \text{ W/m}^2\text{K}$ to $2000 \text{ W/m}^2\text{K}$, slightly less than
310 expected values, obtained from previous standard empirical correlations.

311 **5. NOMENCLATURE**312 **Bi** Boiling number313 d_h Hydraulic diameter, m314 h Heat transfer coefficient, $W/m^2 K$ 315 h_L Heat transfer coefficient for the liquid phase, $W/m^2 K$ 316 k_L Liquid thermal conductivity, W/mK 317 q'' Heat flux, W/m^2 318 Q_{max} Maximum power supplied to the fluid, W319 Re_{LO} Reynolds number for total flow assumed as liquid320 T_{win} Internal temperature of the heated wall, K321 T_{sat} Saturation temperature, K322 We_{LO} Weber number for total flow assumed as liquid323 x Vapour quality324 X Martinelli parameter325 ΔT_{sh} Wall superheat, K326 ρ_L Liquid density, kg/m^3 327 ρ_V Vapor density, kg/m^3 328 **ACKNOWLEDGEMENTS**

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